

School of

WRITTEN EXAMINATION

Course Biochemi	stry	
Examination Sup	ervised w	ritten examination
Course code Ke314G Credits for written examination 3		
Date 20240302		Examination time 9.15-12.30
Examination resp	oonsible P	atric Nilsson/Mikael Ejdebäck
Teachers concern	ied	
Aid at the exam/a	appendice	es; calculators
Other All answers	s must be	given in the exam sheet. Answers given in additional sheets will NOT be
considered.		
Instructions		Take a new sheet of paper for each teacher.
		Take a new sheet of paper when starting a new question.
	\boxtimes	Write only on one side of the paper.
	\bowtie	Write your name and personal ID No. on all pages you hand in.
	\bowtie	Use page numbering.
	\boxtimes	Don´t use a red pen.
	\boxtimes	Mark answered questions with a cross on the cover sheet.
Grade points: to p	pass the ex	xam, all learning objectives require the grade E or higher. To pass a learning
objective, 50% co	rrect ansv	wers are required.
F<50%<=E<60%	5<=D<70	%<=C<80<=B<90%<=A

Examination results should be made public within 18 working days $Good\ luck!$

Total number of pages

Supervised written examination (Biochemistry)

Course code: KE314G

Important information regarding the exam:

The supervised written exam examines two objectives in total:

- -describe the metabolism of carbohydrates, proteins and fats, and its regulation and integration, (10p in total)
- be able to describe the role of proteins as catalysts and perform enzyme kinetic analyses and calculations (40p in total)

To pass the supervised written exam, all learning objectives require the grade E or higher. To pass a learning objective, at least 50 % correct answers are required.

Important things to keep in mind while writing the exam: The teacher who corrects the exam is not a mind-reader. This means that you need to show every step in your calculations otherwise it is very difficult or even impossible to follow your line of thinking. In the end, this will make a huge difference in the number of points you get on a question if you, by chance, make a simple mistake. It is strongly recommended that you make a flow-chart with all steps required to solve a question before you jump into your calculations.

All answers and calculations should be given in this exam sheet. No additional or extra sheets are allowed. Answers given on an extra sheet will not be considered.

Most importantly, believe in yourself. There are no surprises in this exam. We have talked about all the things over and over again.

Good luck

Patric and Mikael

	Learning objective: be able to describe the role of proteins as catalysts			
	and perform enzyme kinetic analyses and calculations. To pass the learning objective, 50% correct answers are required. (5 out of 10p is required)			
1.	In the figure below, a hy	pothetical Line-weaver B	urk plot is depicted. In	
	the figure below draw an	d explain how		
	a) competitive inhibi			2p
	b) non-competitive in			2p
	c) uncompetitive inh			2p 2p
	c) uncompetitive iiii	15/10/		
	would affect K_M and V_{max}			
		A		
	$\frac{1}{V}$			
	V			
	-1			
	$x intercept = \frac{-1}{K_M}$		ition	
	7	No inhit	, ·	
	/	No		
	/			
			→	
			1	
			$\overline{[S]}$	
			[၁]	
		y intercept =	: ——	
		J	V_{max}	
2.	Enzymes can be divided i	nto classes depending on	what reaction they	
	catalyse. Complete the to	able below for four enzyr	ne classes	
	Enzyme class	Reaction catalysed	Example	0.5
	Oxidoreductases			0.5p
	Lludrahuaga			0.5p
	Hydrolyases			
	Transferases			0.5p
				0.06
	Lyases			
1 !	-1			0.5p
1				

3.		_
	Explain, in detail, how the pyruvate dehydrogenase complex converts 1 molecule of pyruvate to one molecule of acetyl-CoA (Hint: the enzyme	2р
	consists of 3 subunits)	
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Lea	rning objective: describe the metabolism of carbohydrates, proteins and fat	s, and
1	regulation and integration. To pass the learning objective, 50% correct answe	rs
	required. (20 out of 40p is required)	
1		
4.	The citric acid cycle is one of the most metabolic pathways in our cells.	6p
4.	a) Explain the Krebs cycle in detail and highlight where NADH,	6р
4.		6р
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	b) How many molecules of CO2, ATP, FADH2, and NADH are generated from 1 glucose molecule	2р
	c) What is the role of the malate-aspartate shuffle and glycerol-3- phossphate shuffle	1p
5.	Name six electron carriers in the electron transport chain	3p
6.	Fatty acid synthesis a) Which enzyme is responsible for the conversion of Acetyl CoA to Malonyl-CoA	1p
	b) Explain the principles of Fatty acid synthesis	6р

	c) What is the role of NADPH in fatty acid synthesis and how is it generated?	4p
7	a) Explain the four steps involved in beta oxidation	6p

	b) Why is it called beta oxidation?	1р
	c) How many molecules of acetyl-CoA, NADH and FADH2 and ATP are produced when an 18-carbon fatty acid undergoes beta oxidation	4p
8	Explain how NH3 or NH4 [†] is neutralized or converted into Urea in the Urea-cycle	3р

9	Explain how we adopt to endurance training at the physiological, cellular and molecular level	3р
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